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Blue–green luminescence from Carbon doped Silicon–Rich Silicon Oxide

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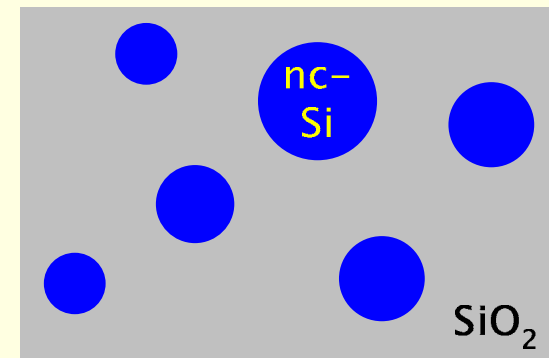
- Introduction: Si photonics
- Experimental details
- C doping effect on visible luminescence of silicon rich silicon oxide
- Tb luminescence from Tb and C co-doped silicon rich silicon oxide
- Implication for Si micro photonics
- Summary and conclusion

Silicon as photonic materials

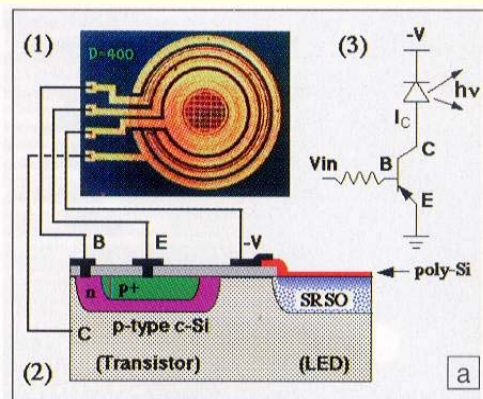
- Very faithful technology and process in micro- or nano- scale
 - Important related compound for photonics: Si, SiO₂, SiC and Si₃N₄
 - 'Silicon' is thus promising candidate for micro-photonics
- And various successful Si photonic devices have been demonstrated (e.g, modulator, MEM optic-switch, and Photonic bandgap)

Silicon-rich silicon Oxide (SRSO)

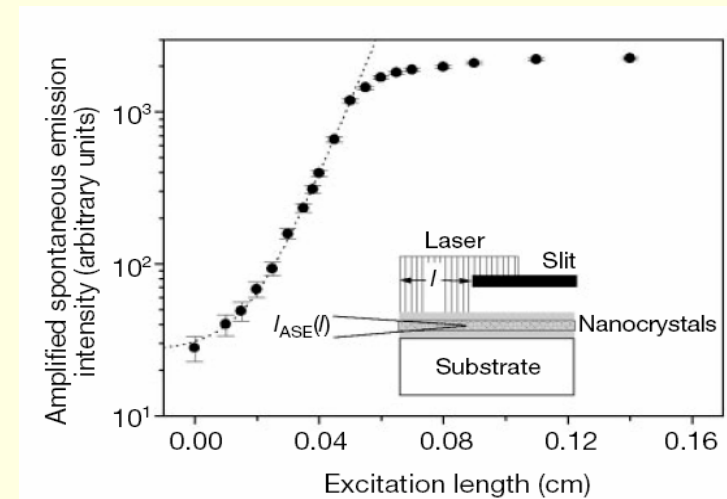
- Nanocluster Si (nc-Si) embedded in SiO₂ matrix
 - Quantum confinement effect of nc-Si: increases both oscillator strength and bandgap
- Efficient visible luminescence at room-temperature
- Easy for formation of dense and robust nc-Si + standard Si process technology
- Promising candidate for Si microphotonics



State of Art of Si photonics using nc-Si



Nc-Si LED integrated with electric circuit (Hirchman, Nature 1996)



Optical amplification (Pavesi, Nature 2000)

However, several limitations still have to be overcome

Features of nc-Si: Limitation for display application

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Generally, not visible for eyes

- At most cases weak Red—near infrared luminescence

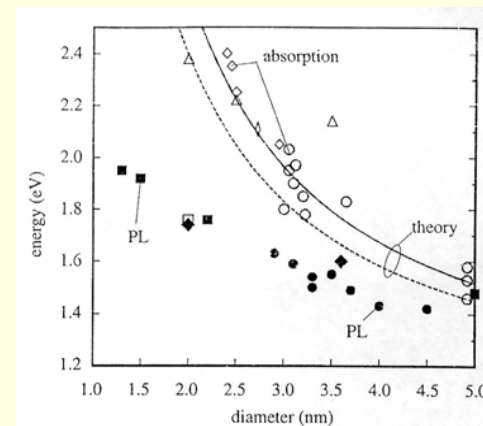
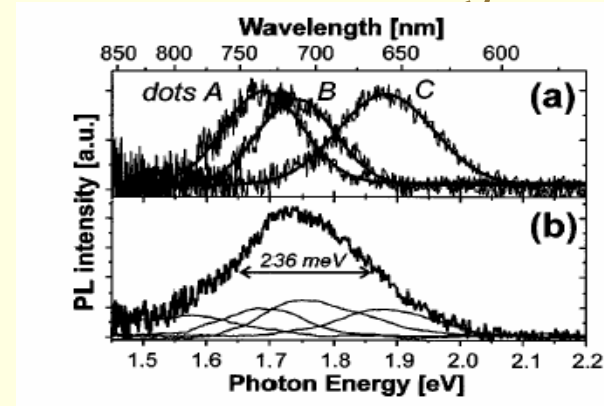
(Valneta, APL 2002)→

Low optical transition energy from even 2—3 nm sized nc-Si

- while bandgap can exceeds 3 eV, PL peak is limited under 2 eV

(Lanoo, PRB)→

Too broad for even single nc-Si



Strong and sharp visible luminescence from SRSO: HOW To?

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How to induce strong and sharp visible luminescence from SRSO?

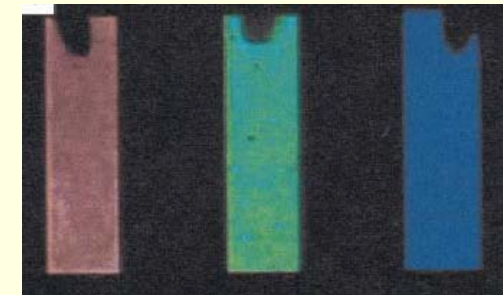
1st HINT.

Rare-earth doped wide bandgap material (e.g., GaN)

If bandgap is wide enough to excite incorporated RE ions

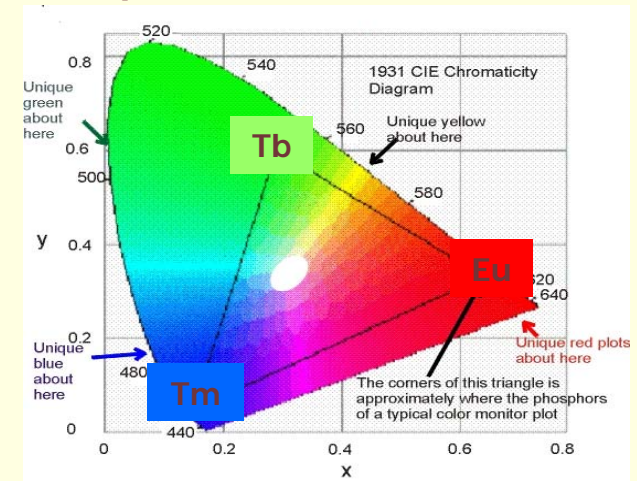
→ Sharp visible 4f transitions of RE ions are feasible

→ Some visible 4f transitions of RE ions: correspond to primary color standard for RGB (Red:Eu or Pr, Green:Er or Tb, Blue:Tm)



Eu Er Tm

Electroluminescences of RE doped GaN (Steckl *et al.*)



CIE chromacity diagram

Solution 1. Rare earth doping into SRSO

Surface states of nc-Si: determines optical transition

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2nd HINT

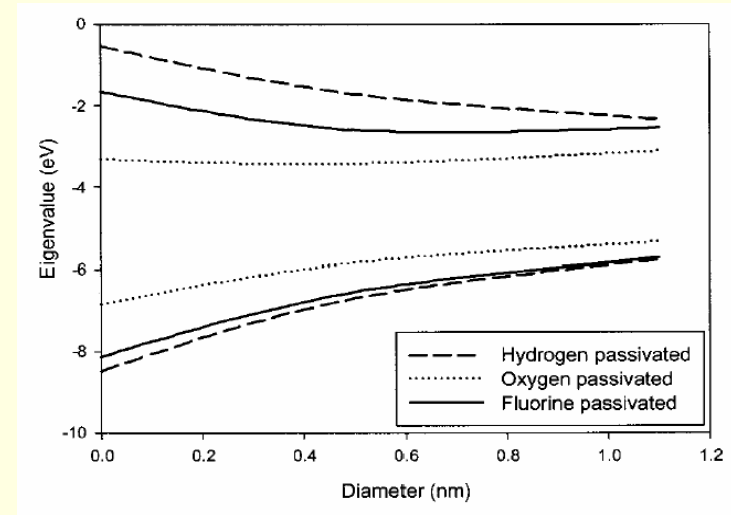
The effect of passivation on the optical transition of nc-Si

Si=O bonds in SRSO: responsible for strong Stokes shift

Species of nc-Si surface passivation
→ Strongly modify electronic states and optical transition



Solution 2. Suitable impurity incorporation into SRSO for the increase of optical transition energy



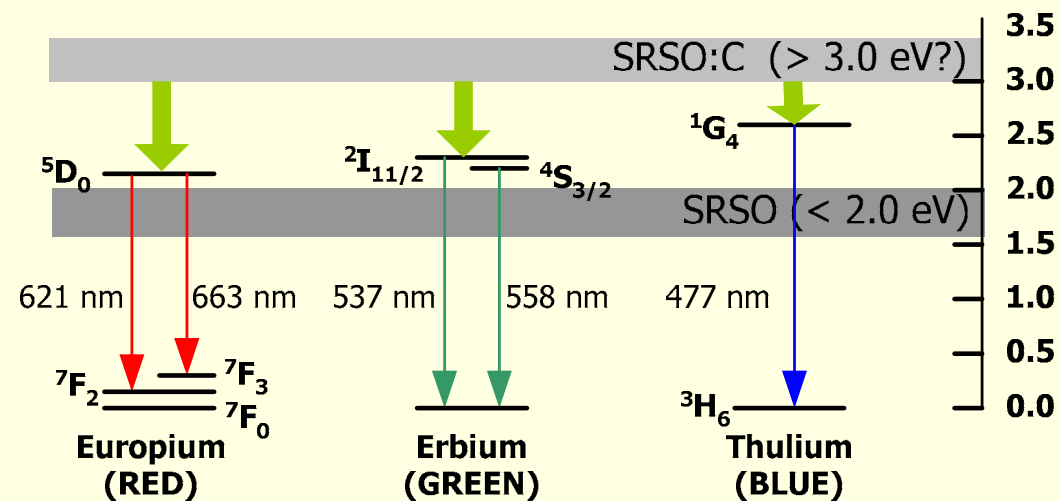
(Puzder, JCP)

Blue-white luminescence of SRSO:C and its utilization for visible phosphors

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Incorporation of suitable impurities to SRSO film

- Can modify surface states of nc-Si
- Can further enhance both optical gap and luminescence efficiency
- Can be utilized as wideband gap material for visible phosphors (such as rare earth ions)
- Finally, strong sharp visible luminescence may be feasible from nc-Si



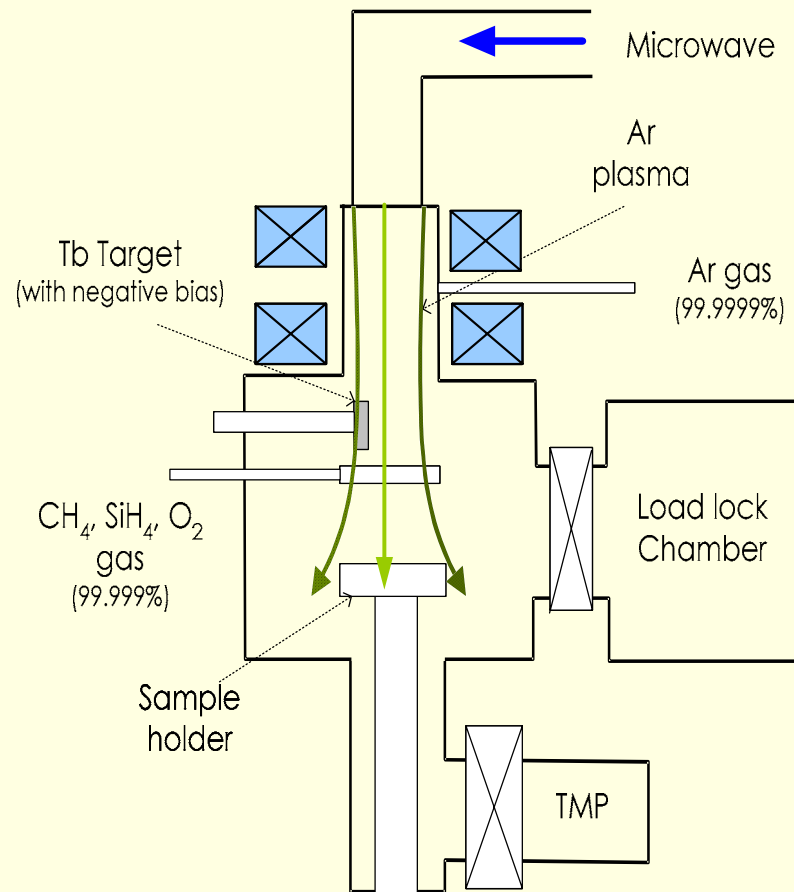
→ NOW, we propose C dope doped SRSO (SRSO:C) as host matrix for sharp and visible 4f transition of incorporated RE ions

Film fabrication

- **Deposition method:** ECR-PECVD
- **Gases:** SiH_4 , O_2 (fixed) & CH_4 (varied)
- **Tb doping:** concurrent sputtering of solid target

Film Characterization

- Rutherford backscattering Spectroscopy
- Wavelength dispersion X-ray spectroscopy
- FTIR & SEM

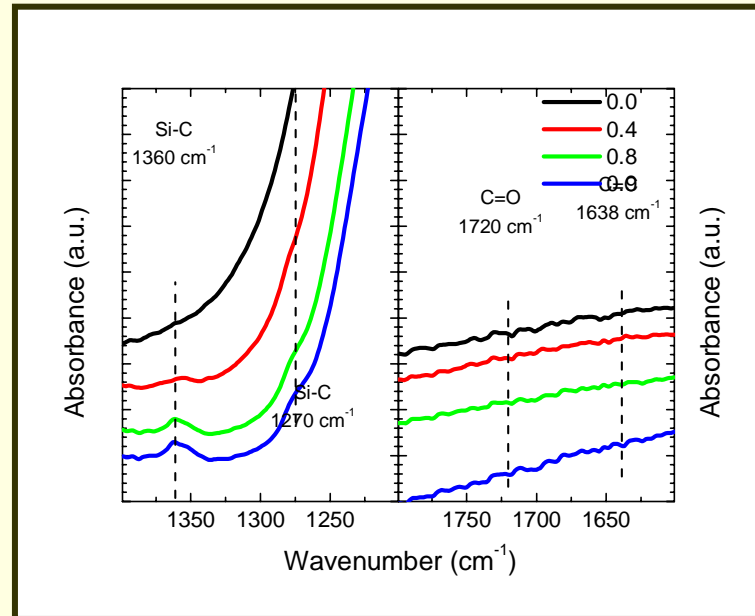


Post treatment

- **Thermal Annealing:** 700—1250 °C for 30 min under Ar environment
- **Hydrogenation:** thermal anneal @700 °C for 1h flowing H₂ forming gas

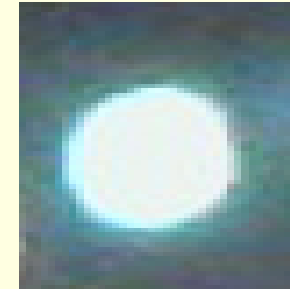
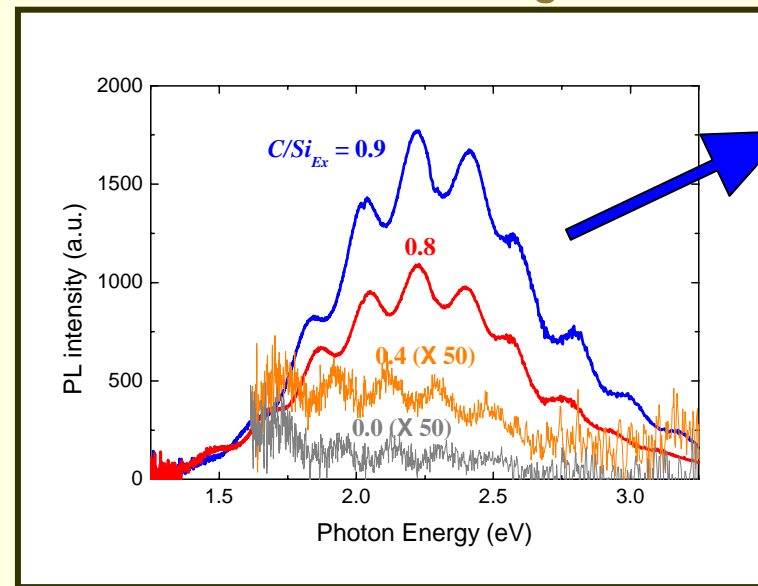
Photoluminescence Measurement

- **Pump Source:** 325 nm of HeCd laser
- **Detectors:** InGaAs(Cs) photomultiplier tube or Charge coupled device
- **Luminescence decay traces:** recorded with digitizing oscilloscope using mechanical chopper



- Films were annealed for 5min @ 950 °C under Ar environment
- Confirmed Si-C bonds (1360 and 1270 cm⁻¹)
- NO C=O (1720 cm⁻¹) and C=C (1638 cm⁻¹) bonds
- Incorporated C ions are bonded only with Si
- Excluded the possibility of the presence of silicon oxycarbide phase

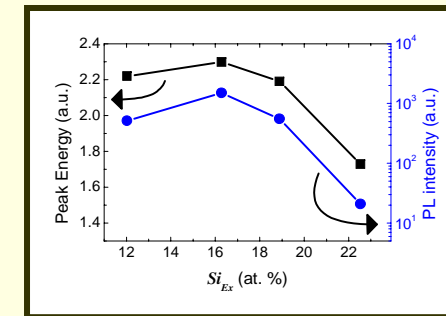
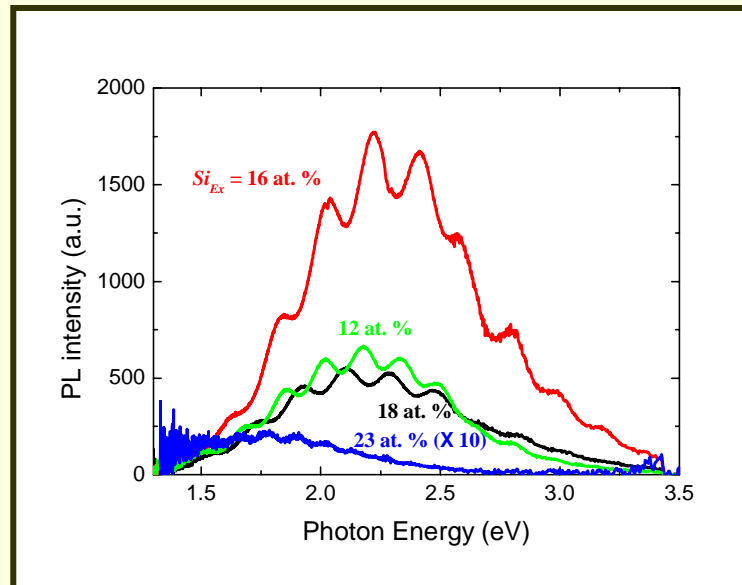
Camera image →



- Films were annealed for 5min @ 950 °C under Ar environment
- Sigmoid oscillation in PL spectra: due to internal reflection
- Very little PL without C, and ~ 500 folds increase of PL by C doping
- Intense blue-white luminescence, visible to naked eyes, under ambient condition

The effect of excess Si content on visible luminescence of SRSO:C

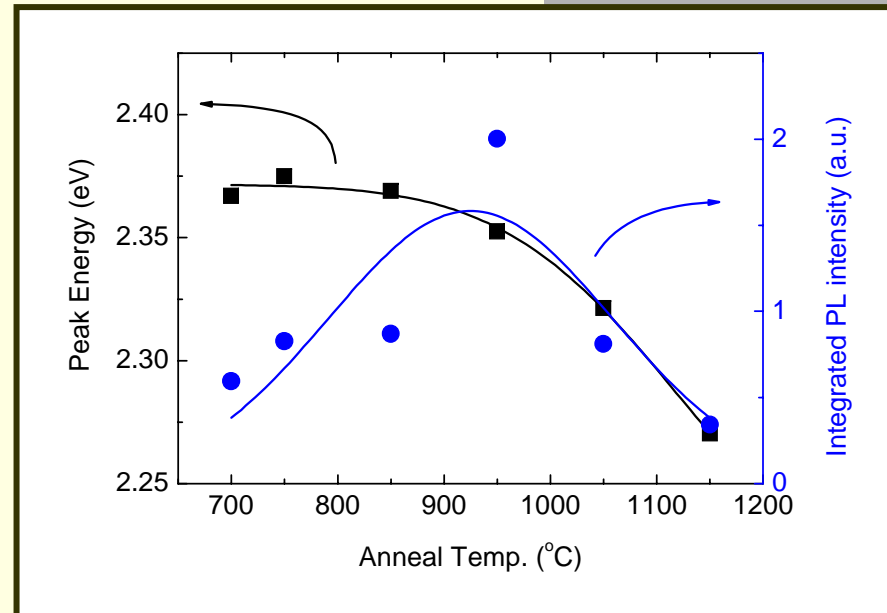
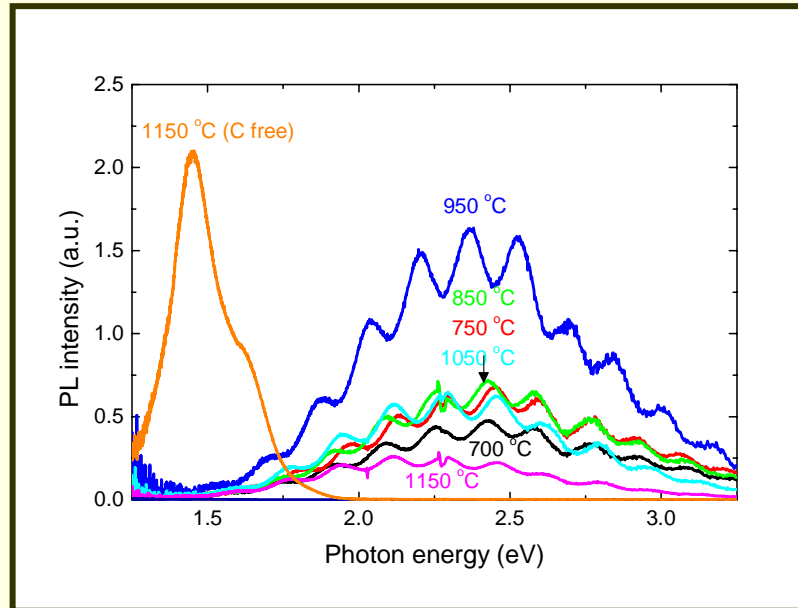
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- The C to excess Si ratio is almost constant (near unity)
- The increase of C and excess Si
 - Strong red shift from 2.3 to 1.7 eV
 - Reduction of PL intensity by 100 times
- Red Shift of PL peak : consistent with C free SRSO

The effect of anneal temperature on visible luminescence of SRSO:C

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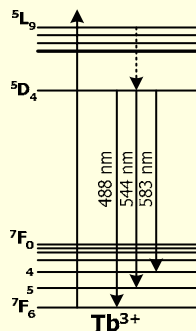
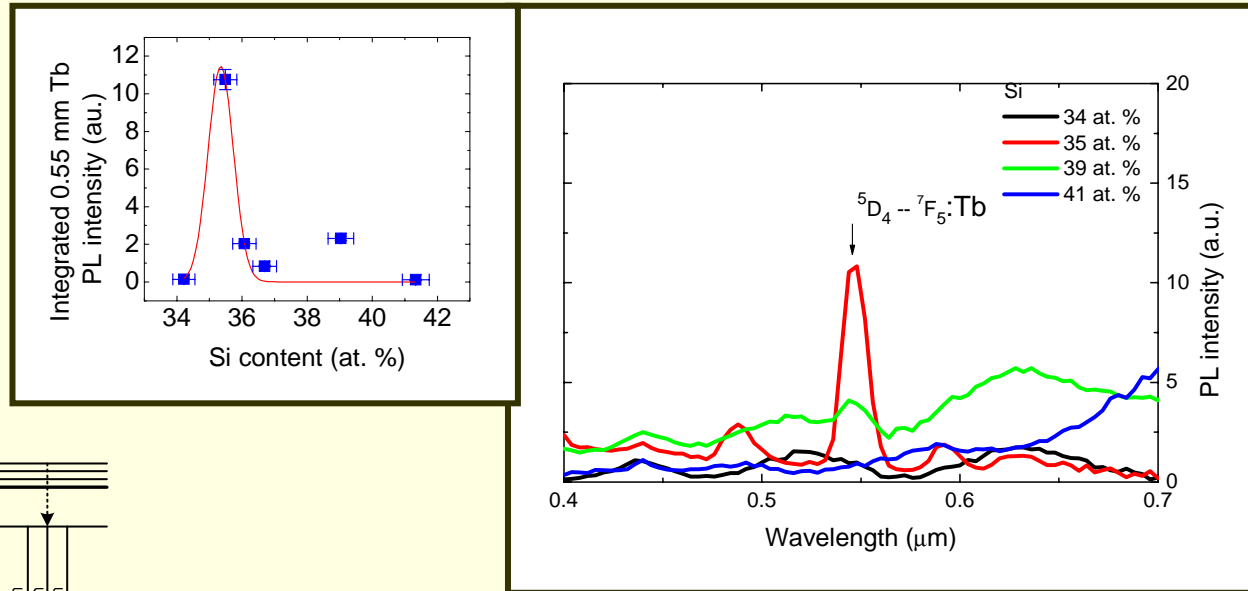
- Si content: 39 at. % with C to excess Si ratio of ~ 0.9
- All films were annealed and hydrogenated
- PL intensity of SRSO:C is larger than that of SRSO by factor of 3 (between optimum films)
- The increase of C and excess Si
 - ➔ Strong red shift from 2.46 to 2.25 eV
- Red shift of PL peak by increase of anneal temperature: consistent with C free SRSO

The origin of luminescence

- Exact origin is not clear
- C is essential for blue-white luminescence
- Overall trend can be consistently explained with those of nc-Si
 - SiO₂ like matrix embedded with nc-Si which incorporate a high concentration of C
 - Luminescence due to excitons being confined in C-doped nc-Si

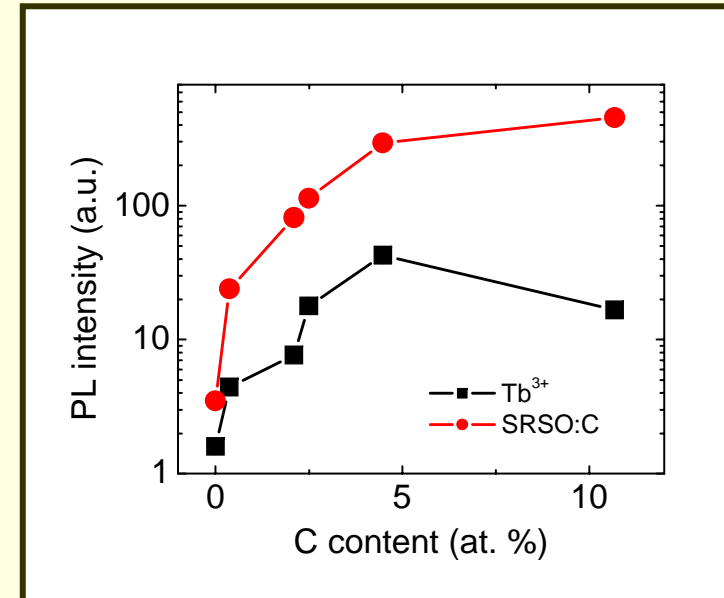
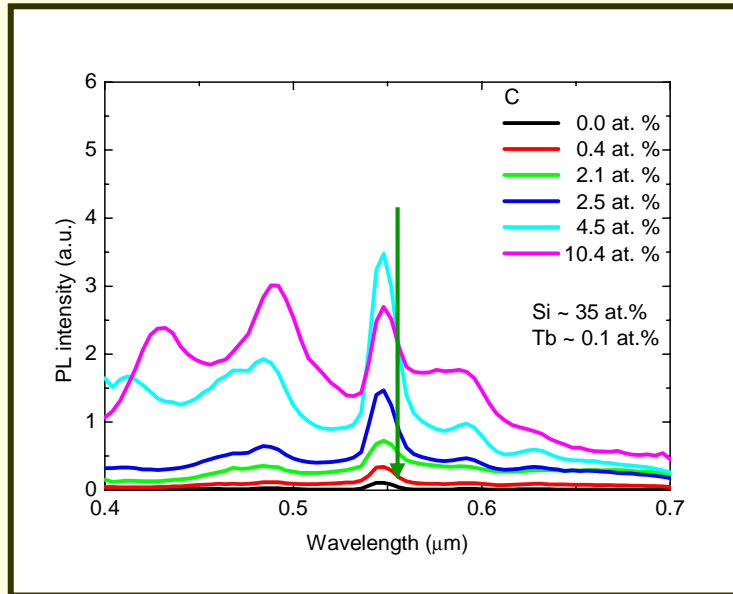
Now check the feasibility of visible RE luminescence!!!

Firstly tried with Tb for green ($\lambda \sim 550$ nm)

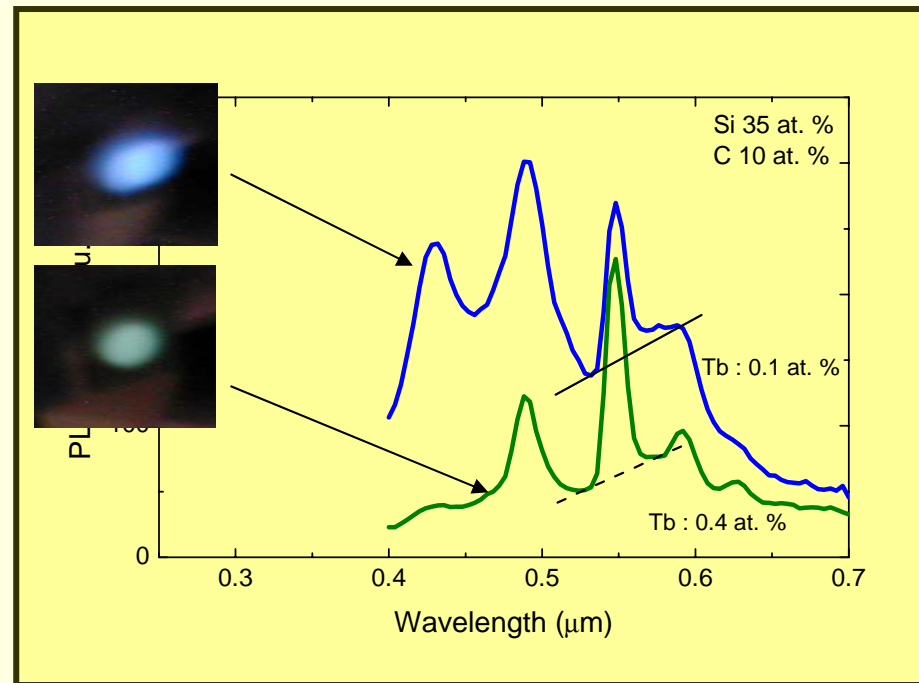


← Energy diagram of 4f transition of Tb³⁺ ions

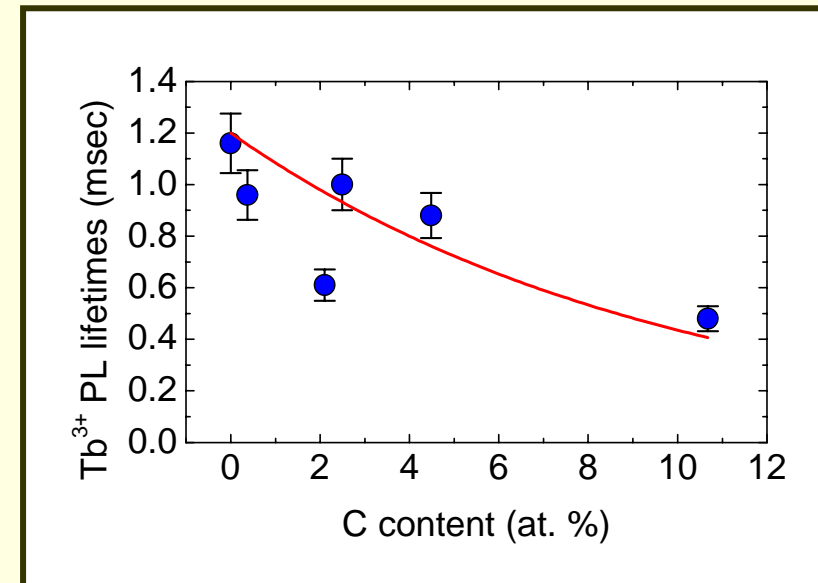
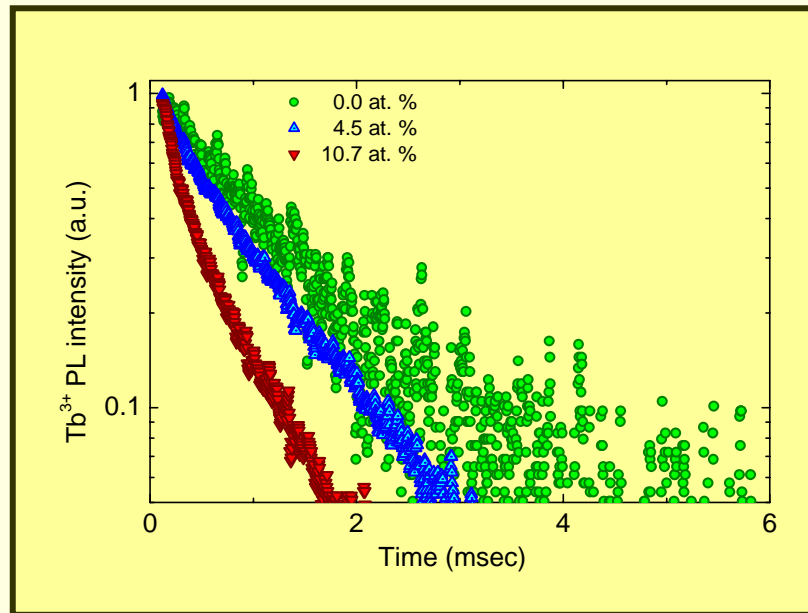
- Weak background luminescence: maybe due to oxide defects
- Clear 543 nm PL due to ⁵D₄ → ⁷F₅ 4f transition of Tb³⁺ within a narrow range of Si content (35– 37 at. %)



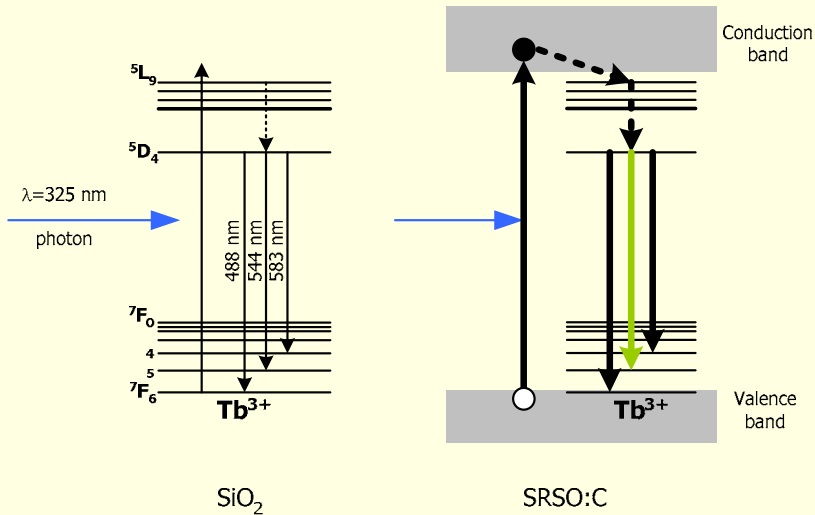
- Broad background luminescence (400-700 nm) due to oxide SRSO:C
- As C content increases both SRSOC and Tb^{3+} luminescence increase
- Tb^{3+} PL is enhanced more than 30 times by C co-doping!!!



- For low Tb content, bluish white and greenish white for high Tb content
- As Tb content increases
 - ➔ Suppression of SRSO:C PL
 - ➔ Increase of Tb^{3+} PL



- SRSO:C host luminescence decay is faster than system response (~10 usec) and is excluded
- As C content increases, Tb³⁺ PL lifetimes are decreased from 1.2 to 0.5 msec
 - ➔ Comparable values with those (~ msec) from Tb-doped SiO₂
 - ➔ Luminescent Tb ions are in SiO₂ matrix



Evidence of energy transfer from SRSO:C to Tb^{3+} ions

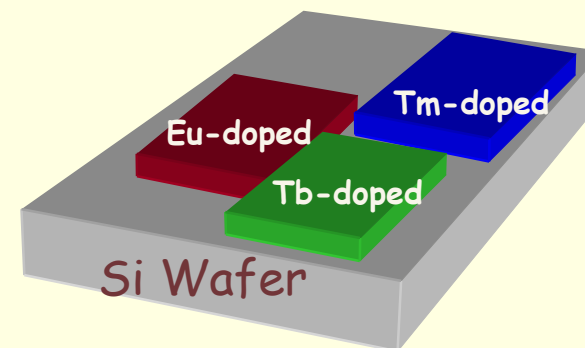
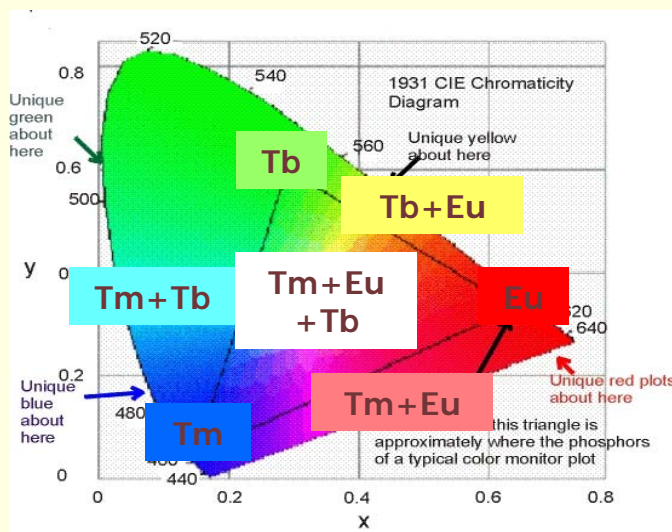
- 325 nm cannot be directly absorbed by Tb^{3+} ions
- Suppression of PL from host matrix by increase of Tb concentration
- Very weak Tb^{3+} PL from Tb-doped silica and narrow range of Si content which can excite Tb^{3+} ions

Role of C co-doping on Tb^{3+} luminescence in Tb-doped SRSO:C

- Increase of the number of high-energy excitons that can excite Tb^{3+}
- ➔ Enhances exciton-mediated excitation rate of Tb^{3+}

Implication: Full color display based on RE-doped SRSO:C

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Feasibility of full color by selection/co-incorporation of RE ions

Integrated RGB display pixel on Si wafer using Si compatible technology

- **Co-doping of C significantly increases blue-white luminescence**
→ The recombination of excitons in C rich nc-Si is likely to be origin of visible luminescence
- **Strong sharp green luminescence can be observed from SRSO by co-doping of C and Tb**
→ Efficient energy transfer from high energy excitons to Tb ions leads to efficient Tb³⁺ luminescence in SRSO:C
- **The feasibility of Si based full color display using RE-doped SRSO:C films**